## An Enantiospecific Route to (6R)-(-)-Massoialactone and (4R,6R)-(+)-4-Hydroxy-6-pentylvalerolactone

Seiichi Takano,\* Masaki Setoh, and Kunio Ogasawara

Pharmaceutical Institute, Tohoku University, Aobayama, Sendai 980, Japan

(Received 9 March 1992)

Abstract: An enantiospecific route to two  $\delta$ -lactone natural products, (6R)-(-)-massoialactone and (4R,6R)-(+)-4-hydroxy-6-pentylvalerolactone, has been developed by using (R)epichlorohydrin as a chiral starting material

The *trans*- $\beta$ -hydroxy- $\delta$ -lactone system is an essential structural feature associated with the HMG-CoA reductase inhibiting activity of compactin and mevinolin.<sup>1</sup> In connection with our synthetic project toward compactin and the related compounds,<sup>2</sup> we report herewith enantiospecific syntheses of two naturally occurring  $\delta$ -lactones, (6*R*)-(-)-massoialactone<sup>3,4</sup> (1) and (4*R*,6*R*)-(+)-4-hydroxy-6-pentylvalerolactone<sup>4,5</sup> (2), from (*R*)-epichlorohydrin (3) by employing our own strategy<sup>2,6</sup> for the construction of *trans*- $\beta$ -hydroxy- $\delta$ -lactone system.



Fig. 1

Treatment of (*R*)-epichlorohydrin<sup>7</sup> (3) with lithium dibutylcuprate, prepared from *n*-butyllithium and copper(I) iodide in the same flask, afforded the chlorohydrin<sup>8</sup> 4,  $[\alpha]_D^{30} - 1.47$  (*c* 1.0, CHCl<sub>3</sub>), in 73.7% yield. Upon exposure to an excess amount (4 equiv.) of lithium acetylide ethylenediamine complex in DMSO solution at room temperature 4 furnished the  $\beta$ -hydroxyacetylene 6,  $[\alpha]_D^{28} + 22.2$  (*c* 1.0, MeOH), in 94.8% yield via a transient formation of the epoxide (5). Since the epoxide (5) was found to be very elusive owing to its high





533

## S. TAKANO et al.

volatility, this one-flask procedure was essential for the efficient conversion. In order to introduce the carbomethoxy group, the terminal acetylene 6 so obtained was then stirred with a catalytic amount of palladium(II) chloride (6 mol %) and copper(II) chloride (2 equiv.) in methanol in the presence of sodium acetate (2 equiv.) under atmospheric pressure of carbon monoxide<sup>9</sup> which gave rise to the expected propiolate ester (7),  $[\alpha]_D^{29} + 11.95$  (c 1.8, CHCl<sub>3</sub>), in 85.1% yield. Partial hydrogenation of 7 followed by acid-catalyzed cyclization of the resulting (Z)-ester (8) afforded (6R)-(-)-massoialactone (1),  $[\alpha]_D^{29} - 107.52$  (c 1.07, CHCl<sub>3</sub>) [lit.<sup>4b</sup>  $[\alpha]_D - 109$  (c 9.1, CHCl<sub>3</sub>)], in 75.7% yield.

When optically active massoialactone (1) thus obtained was exposed to alkaline hydrogen peroxide,<sup>2,6</sup> highly diastereoselective epoxidation occurred to give the epoxide (9),  $[\alpha]_D^{25}$  +81.3 (c 1.0, CHCl<sub>3</sub>), in 83% yield<sup>10</sup> as a single product presumably via a stereoelectronically favored transition state<sup>11</sup> (1a). Subsequent treatment of 9 with the phenylselenolate complex,<sup>2,6,12</sup> generated from diphenyl diselenide and sodium borohydride in the same flask, allowed facile and regioselective cleavage of the epoxide bond to give the desired natural product, (4R,6R)•(+)-4-hydroxy-6-pentylvalerolactone (2),  $[\alpha]_D^{28}$  +38.4 (c 1.15, CHCl<sub>3</sub>)<sup>13</sup> [lit.<sup>5a</sup>  $[\alpha]_D^{25}$  +27.4 (c 11.7, CHCl<sub>3</sub>)], in 74.2% yield.



Scheme 2

## **References and Notes**

- 1. For reviews, see: Endo, A. J. Med. Chem. 1985, 28, 401; Vega, L.; Grundy, S. J. Med. Assoc. 1987, 257, 33.
- 2. Takano, S.; Shimazalki, Y.; Sekiguchi, Y.; Ogasawara, K. Synthesis 1989, 539.
- Determination, see: (a) Cavill, G. W. K.; Clark, D. V.; Whitfield, F. B. Aust. J. Chem. 1968, 21, 2819. (b) Mori, K. Agric. Biol. Chem. 1976, 40, 1617.
- Chiral syntheses, see.<sup>1</sup>; (a) Ref. 3a. (b) Pirkle, W. H.; Adams, P. E. J. Org. Chem. 1980, 45, 4117. (c) Bennett, F.; Knight, D. W. Heterocycles 1989, 29, 639. (d) Asaoka, M.; Hayashibe, S.; Sonoda, S.; Takei, H. Tetrahedron Lett. 1990, 31, 4761. (e) Bennett, F.; Knight, D. W.; Fenton, G. J. Chem. Soc. Perkin Trans. 1 1991, 1543. (f) Romeyke, Y.; Keller, M.; Kluge, H.; Grabley, S.; Hammann, P. Tetrahedron 1991, 47, 3335. (g) Bonini, C.; Pucci; P.; Racioppi, R.; Viggiani, L. Tetrahedron:Asymmetry 1992, 3, 29.
- 5. Structure, see: (a) Vesonder, R. F.; Stodola, F. H.; Rohwedder, W. K. Can. J. Biochem. 1972, 50, 363. (b) Ref. 4c and 4e.
- 6. Takano, S.; Shimazaki, Y.; Moriya, M.; Ogasawara, K. Chem. Lett. 1990, 1177: Takano, S.; Shimazaki, Y.; Iwabuchi, Y.; Ogasawara, K. Tetrahedron Lett. 1990, 31, 3619.
- 7. We thank DAISO Cp. Ltd., Japan for donation of a large quantity of (R)-epichlorohydrin ( $\geq$ 98% ee).
- 8. cf. Takano, S.; Yanage, M.; Takahashi, M.; Ogasawara, K. Chem. Lett. 1987, 2017.
- 9. Tsuji, J.; Takahashi, M.; Takahashi, T. Tetrahedron Lett. 1980, 21, 849.
- 10. Since some of the epoxy-lactone (9) was saponified under these conditions, acid workup followed by brief reflux of the crude product in benzene in the presence of a trace of pyridinium *p*-toluenesulfonate was necessitated to revert the contaminated seco-acid into the lactone (2).
- 11. Cf. Deslongchamps, P. 'Stereoelectronic Effects in Organic Chemistry,' Pergamon, Oxford, 1983, p. 209.
- 12. Miyashita, M.; Suzuki, T.; Yoshikoshi, A. Tetrahedron Lett. 1987, 28, 4293: Miyashita, M.; Hoshino, M.; Suzuki, T.; Yoshikoshi, A. Chem. Lett. 1988, 507.
- 13. The synthetic 2 has much higher specific rotation value to the reported value for the natural 2,<sup>5a</sup> but was virtually identical to the corrected [+38.4 <sup>4c</sup>; +37.7 <sup>4e</sup>] calculated by Knight and coworkers based on their asymmetric synthesis of optically enriched 2.<sup>4c</sup>,<sup>e</sup>